



## Scientific Research

## Application of antisolvent precipitation method for encapsulation of date seed extracts of Rabbi variety in zein protein biopolymer

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**ABSTRACT****ARTICLE INFO**

The aim of this study was to investigate the antioxidant properties of date seed methanolic extracts of Rabbi variety and evaluating the physicochemical properties of zein nanoparticles loaded with these extracts. Ultrasonic treatment was used to extract methanolic extract from date seed powder and antisolvent precipitation method was used to encapsulate date seed methanolic extracts in amounts of 0.05, 0.1, 0.15 and 0.2 g in zein biopolymer carrier. In this study, the total phenolic content (TPC) and half maximal inhibitory concentration ( $IC_{50}$ ) value of date seed extracts of Rabbi variety in 45 min, the temperature of 50°C of ultrasonic bath and 70% methanol solvent concentration were obtained respectively 369.47 mg gallic acid equivalents (GAE)/g dry weight and 5.08 µg/m1. By increasing the ratio of encapsulated extract in zein carrier from 0.1 to 0.4, the encapsulation efficiency from 85.90 to 95.19% and the size of zein particles loaded with methanolic extracts from 129.95 to 183.30 nm increased and the zeta potential of nanoparticles decreased from +19.15 to +14.48 mV. The size and zeta potential of extract-free zein particles were determined 109.30 nm and +21.96 mV, respectively. The results of ATR-FTIR analysis indicated that with increase the ratio of methanolic extract encapsulated in the zein carrier, the stretching peak of O-H bond changed and increased from 3292.85 to 3294.85  $\text{cm}^{-1}$ . In Investigating the FE-SEM images, extract-free zein nanoparticles and zein nanoparticles loaded with date seed extracts had semi-spherical morphology. Overall, the hydrophobic nature of zein carrier caused it to bind with the phenolics of Rabbi date seed extract through non-covalent, van der Waals, hydrogen, and hydrophobic interactions. Therefore, it can be used as a strong carrier for the encapsulation of date seed extracts

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## 1- Introduction

In recent years, novel bioactive phytochemicals derived from medicinal plants have been widely welcomed by societies so as to promote people's health [1]. Functional bioactive compounds are generally categorized into two groups including molecules (vitamins, fatty acids, carotenoids, plant sterols, polyphenols, amino acids, peptides, and proteins) and living cells (probiotics) [2]. Various bioactive substances present in plant extracts and essential oils, such as phenolics, have anti-allergy, anti-inflammatory, antimicrobial, antioxidant, and antidiabetic properties [3]. Although that extracts and essential oils as plant volatile metabolites have been widely used in medicine, cosmetics, and food industries due to the presence of these valuable properties. However, their use is still not completely effective due to problems such as low solubility in water, volatility, and low stability in foods. In addition, extracts and essential oils can create unpleasant flavors in foods and have negative effects on the sensory quality of the products. Moreover, they are prone to environmental stresses such as heat, pH, oxygen, and light, owing to containing phenolic compounds [4]. This limits their direct use in food formulations. Developments in nanotechnology have enabled the food industry to surmount these obstacles to the use of plant extracts and essential oils as preservatives with high nutritional value and health-promoting properties. Nanoencapsulation is the encapsulation process of pharmaceuticals and bioactive compounds in a carrier matrix to protect them against biological destruction, extend their shelf-life, prevent their oxidative degradation, modify their diffusion kinetics, biodistribution, and controlled release, as well as improving the sensory and textural properties of food

products. Nanoencapsulation technology has the ability to face the challenges of the food industry in the field of efficient delivery of functional bioactive compounds as well as managing the release of food flavoring components [5]. Size reduction to nanoscale (less than 1000 nm) can play a key role in the bioavailability, solubility, and targeted delivery of nutrient compounds, due to increasing the surface-to-volume ratio of the particles [6].

In nanotechnology, carrier matrices are divided into three groups, namely lipid, polymer, and lipid-polymer. Potential toxicity of polymers, absence of industrial-scale manufacturing processes, limited applications of suitable biopolymers (protein and carbohydrates), and the need for organic solvents have restricted the use of polymer nanocarriers. Despite the countless benefits of protein and carbohydrate nanocarriers, there is no possibility of their industrialization due to the use of extensive complex thermal or chemical processes with no full control capability [7].

Zein, a polyamine biopolymer carrier, is the major protein of corn. It is a hydrophobic and thermoplastic polymer, and owing to its hydrophobicity, this protein is transparent and insoluble in water. It is also resistant to bacteria and inherently a free radical scavenger [8]. Among the unique properties of zein as a carrier are its remarkable resistance to digestive enzymes, low digestibility in the digestive system, and wide applications in the delivery of bioactive compounds by simply encapsulating hydrophobic molecules through strong hydrophobic interactions with them [9].

Nowadays, date seed, one of the major wastes of date fruit, has received considerable attention because of its pharmaceutical properties and comprising a variety of nutritional ingredients including

vitamins, sugars, carbohydrates, dietary fibers, minerals, amino acids, and lipids [10]. It is rich in protein (5.1 g/100 g), fat (9 g/100 g), dietary fibers (73.1 g/100 g), phenolics (3942 mg/100 g), and antioxidants (80400 µmol/100 g) [11]. Date seed has a low sugar content which approximates to 7.2-7.6% [12]. The phenolic acids identified in date seed include gallic acid, protocatechuic acid, vanillic acid, caffeic acid, p-coumaric acid, ferulic acid, m-coumaric acid, and o-coumaric acid [13]. Date seed extracts are efficient in treating cerebral ischemic damages by reducing the brain oxidative stress [14]. The dietary fibers present in the powder and extracts of date seed play important roles in preventing diabetes, obesity, and hyperlipidemia, as well as protecting consumers against hypertension, coronary artery disease, high cholesterol levels, colon cancer, prostate cancer, and intestinal disorders [15].

Rabbi date is one of the native dates of Sistan and Baluchistan province, which is grown in Zabol, Saravan, Iranshahr and Chabahar cities. The color of Rabbi date is dark brown to black. Rabbi date seed is smaller than the pulp and it is easy to wash due to its skin and texture. Rabbi date is classified as a semi-dry date with a moisture content of below 14%. The size of this variety of date is a little tall and its normal length is between 3-5 cm. A limited number of studies have ever been performed on the antioxidant properties of Rabbi date seed extracts. Radfar et al (2019) investigated the TPC and IC<sub>50</sub> of the ethanolic (80%) extracts of Rabbi date seed, prepared through maceration. They found out that the TPC and IC<sub>50</sub> were respectively equal to 2423 mg gallic acid equivalents (GAE)/100 g dry matter and 20.4 µg/ml. They also realized that the phenolic compound profile of Rabbi date seed was composed of gallic acid, vanillic acid, 3,4-dihydroxybenzoic acid, 2,5-dihydroxybenzoic acid, cinnamic acid, caffeic acid, and chlorogenic acid [16].

Furthermore, no research has thus far been conducted on the application of zein to encapsulate date seed extracts, as well as examining the properties of the zein particles loaded with the extracts. At the same time, Jivan, Yarmand, and Madadlou (2014) encapsulated the aqueous extracts of Kabkab date seed using starch nanocrystals in the microemulsion system and evaluated the encapsulation parameters. They declared the mean size, size distribution, and polydispersity index of the starch nanocrystals loaded with the date seed extracts to be 198 nm, 105-376 nm, and 0.162, respectively. The EE of the nanocrystals was equal to 62%, and the morphologies of both the extract-free and extract-loaded nanocrystals were semispherical [17]. Sadeghi, Madadlou, and Yarmand (2014) encapsulated the aqueous extracts of Kabkab date seed through the microemulsification-cold gel formation of whey proteins and assessed the encapsulation parameters. These researchers found the mean size of whey protein isolate after heat treatment, particles without extract and capsules loaded with extract to be 23, 304 and 230 nm, respectively, as well as they reported the partial conversion of the first type amide bond in whey protein isolate to second type amide bond in particles and capsules and the formation of disulfide bands in Fourier transform infrared spectroscopy analysis. In addition, they presented the spherical morphology of most of the particles due to encapsulating without affecting their thermal behavior [18]. Considering the role of nanotechnology in preserving the bioactive compounds of date seed extracts, opening up the possibility of enriching nanofoods, raising the bioavailability, targeted delivery of the bioactive compounds, and increasing their effectiveness in restraining and curing diseases, it has been attempted in this research to investigate the potential of zein, as a biopolymer carrier, for the

encapsulation of Rabbi date seed methanolic extracts.

## 2. MATERIALS AND METHODS

### 2.1 Chemicals and reagents

All the chemicals and reagents used in this study were of analytical grade. Folin-Ciocalteu reagent, gallic acid monohydrate, anhydrous sodium carbonate ( $\text{Na}_2\text{CO}_3$ ), ethanol 96%, methanol 99.9%, and glacial acetic acid were all acquired from Merck Co. Double-distilled water was bought Zolalteb Chemistry Co. 2,2-diphenyl-1-picrylhydrazyl (DPPH) and zein were supplied by Sigma-Aldrich Co.

### 2.2 Date seed powder preparation

The fruits of Rabbi date (*Phoenix dactylifera* L.) were collected from the palm groves of Zabol city (Sistan and Baluchestan province, Iran) in August 2022. After that, the seeds were separated from the pulp and rinsed with tap water to remove the pulp residues. They were then placed on a strainer at ambient temperature for one week to be completely dried. Next, they were ground into a powder under vacuum using an industrial grinder (Flavina Co., Iran) equipped with a cooling system. Eventually, the powder was screened using a sieve with a mesh size of 1 mm.

### 3.2 Ultrasound Assisted Extraction (UAE)

25 g of the date seed powder was dispersed in 100 ml of methanol 70% at a ratio of 1:4 (w/v). The dispersion was thoroughly mixed using a magnetic stirrer at 500 rpm for 30 min. Next, it was poured into a 500 ml capped container to inhibit the vaporization of the solvent during the UAE which was subsequently carried out using an ultrasonic bath (Bandelin, DT 102H, Germany) at 35 kHz, 480 W, and 50°C for 45 min. Following the extraction process, the resulting extract was centrifuged (ATP, RF14000, Germany) at 7800 rpm for 30 min.

Then, it was passed through a Whatman No. 1 filter paper. Afterwards, the solvent was evaporated using a rotary evaporator (Heidolph, Hei-VAP Value Digital, Germany) at 40°C and 200 rpm to prevent the decomposition of the extract phenolics. For evaporating the remaining solvent, the concentrated extract was poured onto a plate and dried in a vacuum oven at 40°C. Finally, the extract was kept at -18°C for further uses [3].

## 4.2 Antioxidant tests

### 4.2.1 TPC

The TPC of the date seed extract was determined through the Folin-Ciocalteu method [3]. For this purpose, an ethanolic (70%) solution of the extract was prepared at 1000 ppm. Then, 100  $\mu\text{l}$  of the solution was poured into a test tube using a sampler. Next, 500  $\mu\text{l}$  of the Folin-Ciocalteu reagent and 1000  $\mu\text{l}$  of distilled water were incorporated into the tube. After 1 min of shaking, 1500  $\mu\text{l}$  of sodium carbonate 20% was added, and the mixture was shaken. The tube was stored in the dark at room temperature for 2 h. Finally, the absorbance value of the solution was read using a spectrophotometer (UNICO, 2150, the US) at 760 nm. The standard curve was achieved by drawing the absorbance values against gallic acid concentrations, and the extract TPC (mg GAE/g) was computed using the curve linear equation:

$$Y = 2.9507 X - 0.039 \quad R^2 = 0.9951$$

Where Y stands for the absorbance value, and X indicates the TPC (mg GAE/ml).

### 4.2.2 Free radical scavenging activity

The DPPH test was conducted to measure the antioxidant activity of the extract [3]. To that end, a methanolic solution of DPPH was prepared at 0.004%, 5 ml of which was further combined with 50  $\mu\text{l}$  of different concentrations (10, 20, 40, 80, 160, and 320 ppm) of the seed methanolic extracts. The solutions were stored in a dark place at

ambient temperature for 30 min. Ultimately, the absorbance values of the samples and the blank were measured spectrophotometrically at 517 nm. The DPPH free radical scavenging activity was quantified by the following equation:

$$\begin{aligned} \text{DPPH scavenging activity (\%)} &= \\ \frac{\text{Blank absorbance value} - \text{Sample absorbance value}}{\text{Blank absorbance value}} \times 100 \end{aligned}$$

## 5.2 Encapsulation

The methanolic extracts of the date seed were encapsulated in zein through the antisolvent precipitation method with some modifications [19]. In order to prepare the solvent phase, 0, 0.05 g, 0.1 g, 0.15 g, and 0.2 g of the methanolic extract powder were dissolved in ethanol 70% and homogenized using a magnetic shaker at 40°C. The volumes of the solutions were made to 15 ml, so their concentrations would reach 0, 3.3, 6.6, 10, and 13.3 mg/ml. After cooling down the solutions to room temperature, 0.5 g of the zein powder was added (zein concentration became 33.3 mg/ml), and the solution was stirred on the magnetic shaker at 25°C for 1.5 h. It was further

## 6.2 Encapsulation tests

### 6.2.1 Encapsulation efficiency (EE)

The amount of the encapsulated extract was determined using the TPC of each powder [20]. To that end, 200 mg of each powder was dispersed in 2 ml of methanol:acetic acid:water mixed at a ratio of 50:8:42 (v/v/v). The dispersion was homogenized on a tube shaker for 1 min and subsequently sonicated in an ultrasonic bath at 480 W, 35 kHz, and 25°C for 20 min. After that, it was centrifuged at 5000 rpm for 10 min. The TPC of the supernatant was quantified through the Folin-Ciocalteu method [3]. EE was calculated using the following equation:

$$\text{EE (\%)} = \frac{\text{Encapsulated TPC}}{\text{Total TPC}} \times 100$$

### 6.2.2 Particle size and zeta potential

homogenized using a digital mechanical homogenizer (IKA, T18 Digital Ultra Turax, Germany) at 12000 rpm for 5 min. The extract-to-zein ratios were equal to 0, 0.1, 0.2, 0.3, and 0.4. To further reduce the particle sizes, the dispersions underwent ultrasonic pretreatment using an ultrasonic homogenizer (Bandelin Sonopuls, HD 2070, Germany) at 20 kHz for 6 cycles (duty cycle = 0.67) in an ice bath. Afterwards, the solvent phase was dropwise added to the antisolvent phase (40 ml of deionized distilled water) at a rate of 1 drop/s under continuous stirring on the magnetic shaker operating at 1500 rpm. With rapid precipitation at ambient temperature, suspensions of zein nanoparticles were obtained, which were immediately centrifuged at 10,000 rpm and 4°C for 20 min. The formed pellets were separated and freeze-dried (Zirbus, VaCo2, Germany) at 0.017 mPa and -57°C for 48 h. They were then ground using an electric grinder and rubbed using a pestle and mortar so as to obtain solid powder nanoparticles. Eventually, the powder samples were kept at 4°C until use.

The size and zeta potential of the nanoparticles were determined through dynamic light scattering (DLS) using a zetasizer (Particulate systems, HD NanoPlus, the US). The samples were dissolved in ethanol 70% (Due to the non-dissolution of zein nanoparticles in distilled water and its high solubility in ethanol solvent with concentrations greater than 60%) at 1% (w/v) and subjected to DLS with a scattering angle of 90° at 657 nm and 25°C [21].

### 6.2.3 ATR-FTIR

The ATR-FTIR spectra of Rabbi date seed methanolic extract and the extract-free and extract-loaded zein nanoparticles were recorded using an ATR-FTIR device (Bruker, TENSOR II, Germany) in the wavenumber range of 4000-500 cm<sup>-1</sup> (16

scans for each samples at a resolution of 4 cm<sup>-1</sup>) [22]. In this method, 5 mg of the nanoparticle sample was completely softened with a pestle and mortar and sprinkled on the ATR diamond crystal surface of the device. The spectra were recorded at 75 psi, and the obtained data were analyzed using the OPUS software.

#### **6.2.4 Nanoparticles microstructure**

The microstructure (shape, morphology, and integrity in the capsule) of the particles was evaluated through FE-SEM (KYKY, EM8000F, China) [23]. For this purpose, the lyophilized powder samples were sprinkled on the silicon plate of the microscope and covered with a layer of gold. The images were captured at an accelerating voltage of 10 kV.

#### **7.2 Statistical analysis**

Mean comparison between the results of the encapsulation tests was carried out using the Tukey's test at 99% confidence level ( $\alpha=1\%$ ) by means of SPSS ver. 28.

### **3. RESULTS AND DISCUSSION**

#### **3.1 TPC**

The mean TPC of the methanolic extracts of Rabbi date seed was found to be  $369.74 \pm 0.85$  mg GAE/g dry matter under the UAE conditions of 45 min, 50°C, and ethanol concentration of 70%. Radfar et al (2019) investigated the TPC of the ethanolic (80%) extracts of Rabbi date seed, prepared through maceration, and understood that it was equal to 2423 mg GAE/100 g dry matter [16]. Dehghanian et al (2017) claimed the TPC of the methanolic extracts of date seed, produced using maceration, to be 175.4 mg GAE/g dry matter [24]. Al-Farsi et al (2007) macerated the seed powders of three Omani dates and realized that their TPCs ranged from 3102 to 4430 mg GAE/100 g fresh weight [25]. The TPCs reported in the

above-mentioned studies are lower, compared with the present research. These differences can highly be due to the effects of the ultrasonic waves on the extraction efficiency of valuable compounds from plant materials in shorter times, as compared to conventional methods. Owing to their high energy content, these waves create shear forces which destroy plant cell walls. As a result, they increase the release of the cell contents into the extraction medium, elevate the mass transfer rate, and reduce the extraction time. Sonication time parameter increases the mass transfer time [3, 26]. Sonication temperature also gives rise to the solubility of polyphenols and accelerates their transfer from plant cells to the extraction medium [27]. Thermal energy improves the extraction efficiency by disrupting the cellular structure of the plant matrix. Destruction of the cell structure causes an increase in the permeability of the cell membrane and the breakdown of secondary metabolites resulting from matrix interactions (polyphenols with lipoproteins) and hence, an increase in the solubility and mass transfer of polyphenols. Additionally, increasing the temperature of the solvent can lower the solvent surface tension and, as a result, elevates the wetting of the plant material, thus enhancing the extraction efficiency. High temperatures decrease the viscosity of the extraction medium; therefore, the solvent penetrates into the plant cell more easily, and the extraction efficiency is enhanced [28]. Because of their high affinity with polyphenols, two-phase solvent systems (alcohol-water) give rise to the extraction efficiency of these substances. Furthermore, solvents with high surface tension and polarity and relatively low vapor pressure and viscosity such as the mixtures of methanol/water facilitate the disintegration of plant cells and the release of the solutes into the extraction medium, thereby increasing the intensity of ultrasonic cavitation and, consequently, the cell

permeability [29]. Babiker et al (2020) performed the UAE of phenolics from roasted and unroasted date seeds with methanol 80% as the solvent at 60°C for 30 min. They reported the TPCs of the seeds to lay in the range of 525.35-595.83 mg GAE/100 g dry matter, which are lower than those obtained in the present study. This difference can be caused by the difference in the date varieties as well as the application of thermal treatment at high temperatures (180-220°C) in roasting the date seeds. The intense roasting of date seed coffee can bring about a decline in its antioxidant potential [30]. Afshari et al (2023) macerated Estameran date seed powder in water and various concentrations of water-ethanol. They maintained that the TPCs of the extracts were equal to 742.37 mg GAE/g dry matter for ethanol 75% and 236.07 mg GAE/g dry matter for water [31]. The TPC of the ethanolic extract was higher than that obtained in the present research. This difference can be attributed to the difference in the date varieties and maturation levels, environment and climate, growing conditions, fertilizers, harvest time, diseases and pests, soil type, processing method, storage conditions, and extraction conditions including solvent type and concentration, time and temperature, and the impact of ultrasound on the extraction efficiency, compared to maceration [32].

### 3.2 IC<sub>50</sub>

IC<sub>50</sub> is defined as the concentration of an extract, which can scavenge 50% of DPPH free radicals. Free radical scavenging activity is defined as the capability of reducing agents like phenolic compounds to transfer hydrogen atoms into free radicals. Assessment of this feature is the most common technique for quantifying an extract antioxidant activity, which is according to the discoloration of the DPPH solution by the extract antioxidants. DPPH is a free

radical stable at room temperature and methanol solvent, and its methanolic solution is purple. If it reacts with an antioxidant, it loses its free-radical characteras, its chain is decomposed; hence, its color is converted from purple into bright yellow due to the donation of a hydrogen atom by the antioxidant to the DPPH molecule. Antioxidants reduce DPPH free radicals to a more stable form. Consequently, the higher the TPC and antioxidant compounds of an extract, the lower its IC<sub>50</sub> [3, 33]. In general, there is a positive correlation between the TPC and antioxidant power of date seed extracts. The mean IC<sub>50</sub> of the methanolic extracts of Rabbi date seed, prepared at the UAE time of 45 min, temperature of 50°C, and solvent concentration of 70%, was equal to 5.08±0.09 µg/ml. Radfar et al (2019) prepared ethanolic (80%) extracts of Rabbi date seed using maceration and found out that their mean IC<sub>50</sub> was equal to 20.4 µg/ml [16], which is higher than that obtained in the present research, revealing the lower antioxidant power of their extracts. Dehghanian et al (2017) reported the IC<sub>50</sub> value of methanolic extracts of date seeds by maceration method equal to 41.30 µg/ml [24]. The IC<sub>50</sub> value is higher than that of the methanolic extract of the present study. Afshari et al (2023) macerated Estameran date seed powder in water and different concentrations of aqueous ethanol and stated that the IC<sub>50</sub> values of the ethanolic (75%) and aqueous extracts were respectively equal to 0.90 and 6.83 mg/ml [31]. The IC<sub>50</sub> value is higher than that of the methanolic extract of the present study. Djaoudene et al (2019) prepared the acetonic (75% acetone and 25% water) extracts of eight Algerian date seeds through maceration in the date maturation stage. They cited that the lowest IC<sub>50</sub> was found to be 37.30 µg/ml [34], which was higher than that of the present research. Ghafoor et al (2022) employed the Soxhlet, supercritical CO<sub>2</sub>, and subcritical CO<sub>2</sub>

methods to extract the antioxidant components of the seeds of Sukari, Ambara, Majdool, and Sagai dates which are native to Saudi Arabia. They declared that the highest antioxidant activity belonged to Majdool date seed extract produced using subcritical CO<sub>2</sub> (IC<sub>50</sub>= 109.69 µg/ml), and the lowest antioxidant activity was associated with Sagai date seed extract prepared through Soxhlet (IC<sub>50</sub>= 353.83 µg/ml). They showed that the IC<sub>50</sub> values of the extracts produced using supercritical and subcritical CO<sub>2</sub> varied between 109.69 and 228.76 µg/ml [35], which are considerably higher than that achieved in the present study. This difference can be brought about to a large extent by the outstanding efficiency of ultrasound in extracting valuable compounds from date seed, compared with conventional extraction methods like Soxhlet and maceration, and novel extraction methods such as supercritical and subcritical CO<sub>2</sub>. It can also be caused by the difference in the genotypes of the date varieties as well as the climatic and environmental conditions of the dates cultivation.

### 3.3 EE

EE normally refers to the exact number of bioactive compounds truly retained in a system [36]. It can be seen in Table 1 that as the concentration of Rabbi date seed methanolic extract was increased from 3.3 to 13.3 mg/ml, or in other words, as the extract-to-zein ratio was elevated from 0.1 to 0.4, EE was raised from 85.90 to 95.19% with 91.42±4.04% being the mean EE. In addition, there was a significant difference in EE of methanolic extracts in zein carrier with various ratios at P<0.01. Calliari et al (2020) investigated the encapsulation of *Hibiscus sabdariffa* extract in zein nanoparticles using the antisolvent precipitation method and declared the EE to be 66.1-100%. They also demonstrated that

the solvent phase composed of more than 1 g of zein powder in 15 ml of ethanol 80%, was able to encapsulate 0.5-1.5 ml of the extract with an EE of above 91% [19], which conforms to that of the present study considering the use of the same carrier and encapsulation method. Sassi et al (2020) applied freeze-drying to encapsulate date seed polyphenols in a wall of egg yolk protein and Gum Arabic and reported the EE varying from 44.06 to 99.75%. They also obtained the lowest EE when the gum was used alone as the wall material, and as the gum-to-protein ratio rose, the EE declined. The enhancing effect of egg yolk protein and other proteins like zein on the EE of date seed phenolics can be attributed to the strong hydrophobic interactions and non-covalent hydrogen bonds between proteins and polyphenols, relative to those between carbohydrates and polyphenols [37]. Liu et al (2022) indicated an increment in the EE of resveratrol in zein-Gum Arabic complex coacervates from 55.85 to 74.20% after sonication. They ascribed this to the steric and non-covalent interactions created between resveratrol, zein, and the gum in more uniformity of nanoparticles and hence, increasing the EE of resveratrol and on the other hand the tremendous thermal, mechanical and cavitation effects of ultrasound waves in increasing the EE of resveratrol [38]. The lower EE obtained in their research compared to ours, is due to the difference in the sonication time as well as using a carbohydrate carrier. In conclusion, the increased EE and the desirable encapsulation of Rabbi date seed methanolic extract in zein nanoparticles with a rise in the extract-to-zein ratio seems to be logical, owing to the creation of strong interactions between the extract polyphenols and the zein nanoparticles, in addition to the boosting effect of ultrasound on improving the EE.

**Table 1.** Encapsulation efficiency of Rabbi date seed methanolic extracts in zein nanoparticles

Sample	Extract-to zein ratio	Encapsulation efficiency (%)
Zein loaded with 0.05 g of date seed extract	0.1	85.90±0.16 <sup>a</sup>
Zein loaded with 0.1 g of date seed extract	0.2	91.13±0.02 <sup>b</sup>
Zein loaded with 0.15 g of date seed extract	0.3	93.49±0.02 <sup>c</sup>
Zein loaded with 0.2 g of date seed extract	0.4	95.19±0.03 <sup>d</sup>

\* All numbers are the average of three replicates.

\*\* There is a significant difference between the values with different letters in each column ( $p<0.01$ ).

### 3.4 Particle size and zeta potential

The size and charge of particles play key roles in the encapsulation and delivery systems of bioactive compounds. Charge or electrostatic attraction/repulsion between nanomaterials is one of the most essential parameters influencing the colloidal stability of nanoformulations and their interactions with the surrounding medium. As a result, zeta potential is a crucial surface parameter in characterizing nanoparticles, which is determined at the relative (diffuse layer) of loosely-bound ions surrounded by the nanoparticle surfaces and free ions in the solution. The ion concentration and pH of the solution impact the composition of the diffuse layer and zeta potential [39]. The mean size and zeta potential of the extract-free zein particles were respectively found to be  $109.30\pm9.75$  nm and  $+21.96\pm0.05$  mV, while those of the extract-loaded ones were equal to  $153.23\pm21.70$  nm and  $+17.21\pm2.04$  mV, respectively. As can be observed in Table 2, with an increase in the extract-to-zein ratio, the nanoparticles enlarged, whereas their zeta potential lowered. The size and zeta potential of the extract-loaded zein nanoparticles lay in the ranges of 129.95-183.30 nm and 14.48-19.15 mV, respectively. Jivan, Yarmand, and Madadlou (2014) encapsulated the aqueous extract of Kabkab date seed in starch nanocrystals

through microemulsification and declared the extract-free and extract-loaded nanocrystal mean sizes to be 164 and 198 nm, respectively [17]. In another research, Sadeghi, Madadlou, and Yarmand (2014) encapsulated the same extract in whey protein using microemulsification-cold gel formation and reported that the mean diameters of the extract-free and extract-loaded protein particles were respectively equal to 304 and 230 nm [17], which are higher than those obtained in the present study. Calliari et al (2020) claimed that the mean diameter of zein particles loaded with ethanolic extracts of *Hibiscus sabdariffa* using the antisolvent precipitation method varied between  $137.9\pm25$  and  $257.4\pm64$  nm. They also observed a slight increment in the particle mean diameters with increasing the zein concentration in the solvent phase from 33 to 100 mg/ml [19]. This is in agreement with our findings in the present research, which can greatly be ascribed to the use of the same method (antisolvent precipitation method) for encapsulating the extracts. At the same time, the lower maximum particle size obtained in the present research, can be due to the dramatic effect of ultrasonic pretreatment on reducing the particle sizes, in addition to the lower concentration of the carrier (33.3 mg/ml) in our research than that (33.3-100 mg/ml) in theirs. The slight

increase in the particle sizes with raising the extract-to-zein ratio in the zein-containing solvent phase is owing to the nucleation and the growth of the particles as a result of the supersaturation created by the antisolvent effect [40]. As the concentration of solutes is elevated, the nucleation rate is increased, too. An excessive increase in the concentration of the solutes leads to the aggregation of the nuclei and the creation of larger particles [19]. Zou et al (2012) maintained that the mean size and zeta potential of zein particles (free of blueberry procyanidins) were  $382.8\pm9.8$  nm and  $+22.80\pm1.56$  mV, respectively, while those of the zein nanoparticles loaded with the procyanidins were in the ranges of  $391.8\pm9.8$ - $447.2\pm24.1$  nm and  $14.24\pm0.89$ - $20.88\pm1.38$  mV, respectively. They also reported the indirect correlation between the zeta potential and size of the procyanidin-loaded zein nanoparticles [41]. Compared with the present research, the mean size and zeta potential of the procyanidin-free zein particles are higher; however, the variation range of the zeta potential and the indirect relationship between the zeta potential and the particle size are consistent with our research. Luque-Alcaraz et al (2022) reported the mean hydrodynamic diameter and zeta potential of extract-free zein particles to be  $159.26\pm5.96$  nm and  $+22.63\pm1.52$  mV, respectively. They also understood that those of the zein particles loaded with the aqueous extract of orange

peel using the antisolvent precipitation method were respectively equal to  $199.96\pm2.87$  nm and  $+11.86\pm0.63$  mV [42]. Compared with our research, the mean size and zeta potential of the extract-free zein particles as well as the mean size of the extract-loaded particles were higher, while the zeta potential of the extract-loaded particles was lower. These differences can be attributed to the differences in the concentrations of zein and the encapsulated extracts. On the other hand, the decrease in the zeta potential with an increase in the extract-to-zein ratio may be owing to the presence of the extract on the particle surfaces. These variations in the surface charge were caused by the hydroxyl groups of the extract polyphenols [42]. The higher the extract TPC, the more its hydroxyl groups (negative charge). This results in a reduction in the positive charge of zein due to the reduced zeta potential. Our results revealed that the zein nanoparticles loaded with Rabbi date seed methanolic extract were positively charged, which can improve the cellular absorption of the encapsulated extract because of the electrostatic interaction between the particle surface and the cell membrane [43]. The enlargement of the particles with increasing the extract-to-zein ratio was brought about by the spatial arrangement of zein amino acid chains interacting with the molecules of the encapsulated extract [44].

**Table 2.** Mean size and zeta potential of extract-free and extract-loaded zein particles

Sample	Extract-to zein ratio	Particle mean size (nm)	Zeta potential (mV)
Extract-free zein	0	$109.30\pm9.75^a$	$+21.96\pm0.05^a$
Zein loaded with 0.05 g of date seed extract	0.1	$129.95\pm3.74^{acd}$	$+19.15\pm0.07^b$
Zein loaded with 0.1 g of date seed extract	0.2	$142.25\pm5.58^{acd}$	$+18.85\pm0.14^b$
Zein loaded with 0.15 g of date seed extract	0.3	$157.45\pm5.86^{bcd}$	$+16.35\pm0.07^c$
Zein loaded with 0.2 g of date seed extract	0.4	$183.30\pm6.93^b$	$+14.48\pm0.05^d$

\* All numbers are the average of three replicates.

\*\* There is a significant difference between the values with different letters in each column ( $p<0.01$ ).

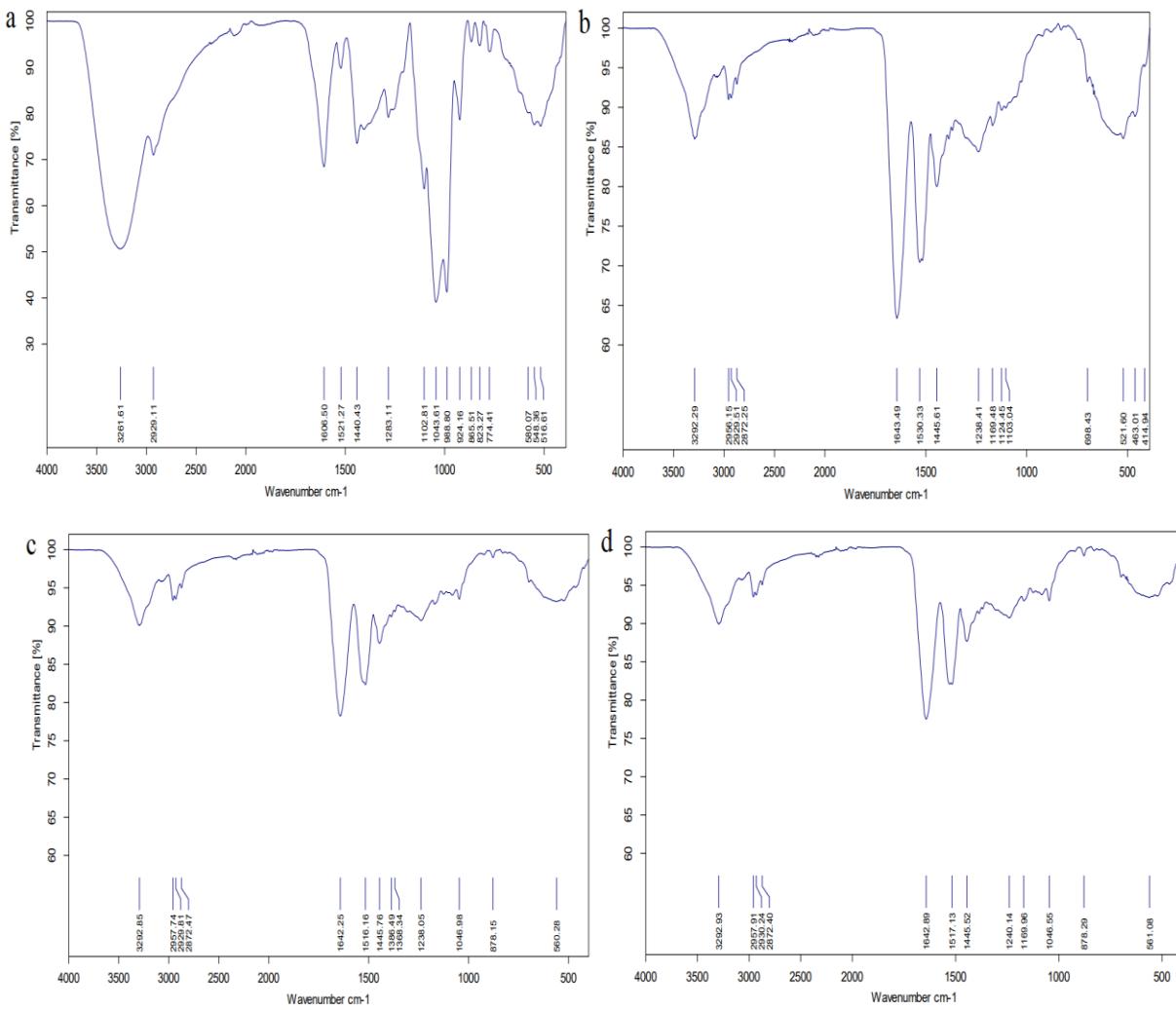
### 3.5 ATR-FTIR

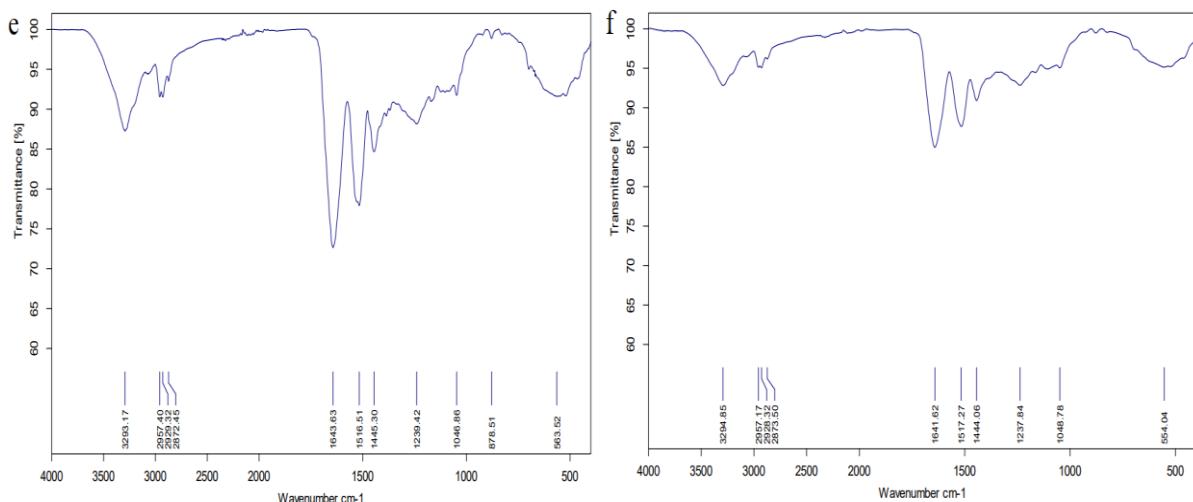
FTIR is a potent method for detecting chemical structure and functional groups in bioactive compounds, and determining their applications with key factors. In this research, ATR-FTIR was utilized to determine the presence of phenolic compounds in Rabbi date seed methanolic extract and their interaction with zein. The spectrum of the pure methanolic extract (Fig. 1(a)) shows a broad band in the wavenumber range of  $3600\text{-}2929\text{ cm}^{-1}$ , characteristic of the stretching vibration of auxochromic hydroxyl (-OH) groups of phenolic compounds [45]. The band at  $3261\text{ cm}^{-1}$  may be associated with the hydroxyl group of the carboxylic group (COOH) on the benzene ring of phenolic acids [46]. The signal ranging from  $2929\text{ to }2820\text{ cm}^{-1}$  is related to the stretching vibration of  $\text{CH}_2$  group of the pure methanolic extract. The band between  $1043\text{ and }988\text{ cm}^{-1}$  pertains to the aromatic groups (C=O) present in the phenolic compounds of pure methanolic extract. These results are in line with those previously presented by González Cruz et al (2022) for the phenolic compounds of Biloxi blueberries extracted with acidic methanol [45]. The ATR-FTIR peaks observed in the spectrum of the pure methanolic extract at  $1606$ ,  $1521$ , and  $1440\text{ cm}^{-1}$  represent the C=C stretch in the aromatic rings of the functional groups of phenolic compounds. The peaks at  $1283$  and  $1043\text{ cm}^{-1}$  are respectively associated with the flavonoids C-O and asymmetrical C-O of the extract polyphenols. The bands at  $774$  and  $516\text{ cm}^{-1}$  are characteristic of the out-of-plane C-H bend present in the benzene rings of polyphenols. These results conform to those of the aqueous extract of Kabkab date seed reported by Bagheri et al (2013) [46]. The ATR-FTIR spectrum of the extract-free zein

nanoparticles (Fig. 1(b)) have signals at  $1643$  and  $1530\text{ cm}^{-1}$  respectively related to the amides type I and II in the amino acid structure of zein protein. The band of amide type I pertains to the axial stretching vibration of C-O, which is located along with the amide type II band with an asymmetrical angulation of N-H bond. The peaks in the range of  $2956\text{-}2872\text{ cm}^{-1}$  indicate the C-H bonds of the  $\text{CH}_2$  and  $\text{CH}_3$  radicals in the zein structure. The band ranging from  $3600$  to  $3014\text{ cm}^{-1}$  belongs to the axial stretch of -OH. These findings almost conform to those previously obtained by Zheng et al (2022) for zein nanoparticles produced through the pH-driven method [47]. The high similarity between the ATR-FTIR spectra of the extract-loaded zein nanoparticles (Fig. 1(c), (d), (e), and (f)) for the extract-to-zein ratios of  $0.1$ ,  $0.2$ ,  $0.3$ , and  $0.4$ , respectively and that of the extract-free one showed that the extract was mainly physically entrapped in the cross-linked nanoparticles, and the probable chemical interactions between the extract and the zein matrix included hydrogen, non-covalent van der Waals, and hydrophobic interactions [17]. It was also realized that as the extract-to-zein ratio was elevated from  $0.1$  to  $0.4$ , the band of O-H...O stretch was shifted from  $3292.85$  to  $3294.85\text{ cm}^{-1}$ , while the spectrum of the extract-free zein particles had an O-H stretch band at  $3292.29\text{ cm}^{-1}$ . The displacement of the -OH band may be due to the formation of strong hydrogen bonds between the amide bonds of zein and the hydroxyl groups of the extract. In other words, the displacement may be caused by the increased activity of hydrogen bonds between the date seed extracts and the zein nanoparticles. Zhang and Han (2018) reported that an increase in the loading amount of rutin into zein-sodium caseinate

nano<sup>+</sup> particles (from 0.05 to 0.2 g), the peak of the stretching vibration of O-H···O was shifted from 3423.89 to 3443.78 cm<sup>-1</sup>. They ascribed this to the more potent activity of hydrogen bonding between the zein-sodium

caseinate nanoparticles loaded with rutin [48]. More compatibility of the zein nanoparticles loaded with Rabbi date seed extracts was achieved owing to hydrogen interactions and hydrophobic forces [46].



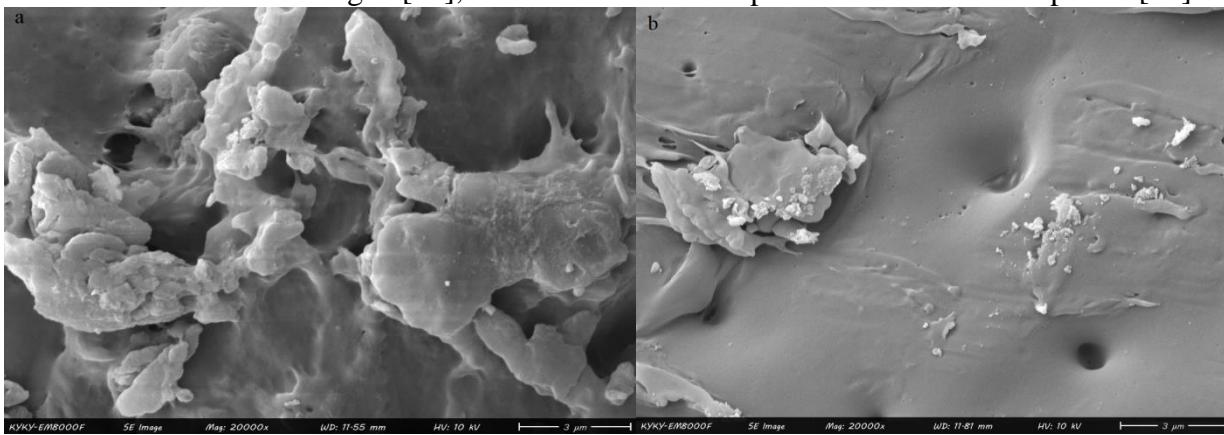


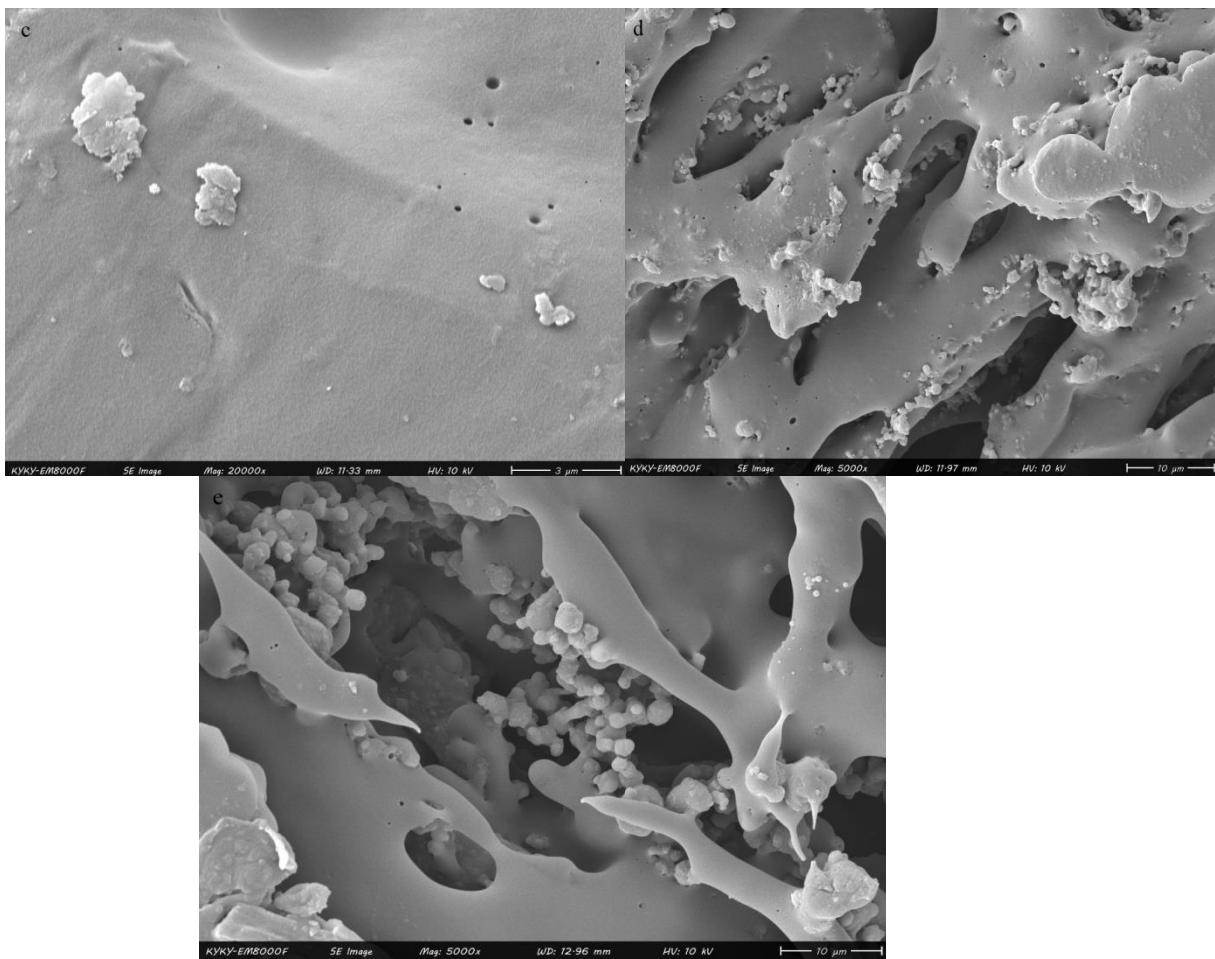
**Figure 1.** ATR-FTIR spectra of Rabbi date seed pure methanolic extract (a), extract-free zein nanoparticles (b), and extract-loaded zein nanoparticles at ratios of 0.1 (c), 0.2 (d), 0.3 (e), and 0.4 (f).

### 3.6 Microstructure

The FE-SEM images of the extract-free and extract-loaded zein nanoparticles are depicted in Fig. 2. It was observed that both the extract-free and extract-loaded nanoparticles had semi-spherical shapes attached to each other and/or overlapped in the dry state [17, 46]. The crosslinks between or the aggregation of the extract-free and extract-loaded zein particles were obvious in the FE-SEM images [49], which

were improved as the extract-to-zein ratio was increased from 0.1 to 0.4. This confirmed the enhanced EE and the increased possibility of the entrapment of the extract in the polymer network of zein with the rise in the extract-to-zein ratio. The elevated number of the extract-loaded zein particles can be attributed to the non-covalent van der Waals, hydrogen, and hydrophobic interactions between the extract and the protein matrix of the capsules [18].





**Figure 2.** FE-SEM images of extract-free zein nanoparticles (a) and extract-loaded zein nanoparticles at ratios of 0.1 (b), 0.2 (c), 0.3 (d), and 0.4 (e).

#### 4. CONCLUSION

Application of non-thermal processes like ultrasound not only raises the extraction efficiency of phenolic and antioxidant compounds of date seed powder, but also reduces the extraction time and retains heat-sensitive compounds, compared to conventional extraction methods. The hydrophobic nature of zein biopolymer caused it to bind with the phenolics of Rabbi date seed extract through non-covalent van der Waals, hydrogen, and hydrophobic interactions. Therefore, it can be a strong carrier for the encapsulation of date seed extracts. As the extract-to-zein ratio was increased, the EE of Rabbi date seed extracts and the size of the extract-loaded zein nanoparticles were raised, whereas particles' zeta potential lowered. The ATR-FTIR results and the FE-SEM images confirmed the excellent potential of zein for encapsulating date seed extracts, creating powerful interactions with hydrophobic core materials, formulating food products, and producing Nano foods.

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## کاربرد روش رسوب ضدحلال در ریزپوشانی عصاره‌های هسته خرما ری بی در بیوپلیمر پروتئین زئین

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### چکیده

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هدف از این مطالعه، بررسی خصوصیات آنتیاکسیدانی عصاره‌های متانولی استخراجی از هسته خرما ری به روش فراصوت و ارزیابی ویژگی‌های فیزیکوشیمیایی نانوذرات زئین بازگذاری شده با این عصاره‌ها بود. تیمار فراصوت جهت استخراج عصاره متانولی از پودر هسته خرما و روش رسوب ضدحلال جهت ریزپوشانی عصاره‌های متانولی هسته خرما به میزان ۰/۰۵، ۰/۱، ۰/۱۵ و ۰/۲ گرم در حامل بیوپلیمری زئین مورد استفاده قرار گرفت. در این مطالعه، میزان ترکیبات فنولی کل و IC<sub>50</sub> عصاره‌های استخراجی از هسته خرما ری در زمان ۴۵ دقیقه، دمای ۵۰ درجه سانتیگراد حمام فراصوت و غلظت حلال متانول ۷۰ درصد به ترتیب ۳۶۹/۷۴ میلی‌گرم اکی والان اسید گالیک بر گرم وزن خشک و ۵/۰۸ میکروگرم بر میلی‌لیتر بدست آمد. با افزایش نسبت عصاره ریزپوشانی شده در حامل زئین از ۰/۱ تا ۰/۴، میزان کارایی ریزپوشانی از ۸۵/۹۰ به ۱۹/۹۵ درصد و اندازه ذرات زئین بازگذاری شده با عصاره‌های متانولی از ۱۲۹/۹۵ به ۱۸۳/۳۰ نانومتر افزایش و میزان پتانسیل زتا نانوذرات از ۱۹/۱۵ به ۱۴/۴۸ + ۱۴/۴۸ میلی‌ولت کاهش یافت. میزان اندازه و پتانسیل زتا ذرات زئین قادر به ترتیب ۱۰۹/۳۰ نانومتر و ۲۱/۹۶ میلی‌ولت تعیین گردید. نتایج آنالیز ATR-FTIR بیانگر آن بود که با افزایش نسبت عصاره متانولی ریزپوشانی شده در حامل زئین، پیک کشش پیوند O-H...O از ۳۲۹۲/۸۵ به ۳۲۹۴/۸۵ cm<sup>-1</sup> تغییر و افزایش یافته است. در بررسی تصاویر FE-SEM، نانوذرات زئین قادر عصاره و بازگذاری شده با عصاره‌های هسته خرما دارای مرفلولوژی نیمه کروی بودند. به طور کلی، ماهیت آبگریز حامل زئین باعث اتصال آن با ترکیبات فنولی عصاره هسته خرما ری از طریق ایجاد ایتراسیون‌های غیرکوالانس، واندروالس، هیدروژنی و هیدروفوب گردیده است. از این رو، زئین می‌تواند به عنوان یک حامل قدرتمند در ریزپوشانی عصاره‌های هسته خرما مورد استفاده قرار گیرد.

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